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JEE MAIN-2022 COMPUTER BASED TEST (CBT)

DATE : 27-06-2022 (MORNING SHIFT) | TIME : (9.00 AM to 12.00 PM)

Duration 3 Hours | Max. Marks : 300

QUESTIONS & SOLUTIONS

PART : PHYSICS

- **1.** A projectile is launched at an angle ' α ' with the horizontal with a velocity 20 ms⁻¹. After 10 s, its inclination with horizontal is ' β '. The value of tan β will be : (g = 10 ms⁻²).
 - (A) $\tan \alpha$ + 5sec α
 - (B) $\tan \alpha 5 \sec \alpha$
 - (C) $2\tan\alpha 5\sec\alpha$
 - (D) $2\tan\alpha + 5\sec\alpha$

Ans. (B)

- **Sol.** $\tan\beta = v_y/v_x = \frac{20\sin\alpha 100}{20\cos\alpha} = \tan\alpha 5 \sec\alpha$
- 2. A girl standing on road holds her umbrella at 45° with the vertical to keep the rain away. If she starts running without umbrella with a speed of $15\sqrt{2}$ kmh⁻¹ the rain drops hit her head vertically. The speed of rain drops with respect to the moving girl is:

V_{rm}

V_{rm}

 $V_{mg} = 15\sqrt{2} \text{ km} / \text{hr}$

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(A) 30 kmh⁻¹

(B)
$$\frac{25}{\sqrt{2}}$$
 kmh⁻²

3 = 45

(C)

Sol.



 \overline{V}_{mg}

V_{rm} = V_{rg}cos 45° V_{mg} = V_{rg}sin 45° = $15\sqrt{2}$ V_{rg} = 30 km/hr ∴ V_{rm} = km/hr

- 3. A silver wire has a mass (0.6 ± 0.006) g, radius (0.5 ± 0.005) mm and length (4 ± 0.04) cm. The maximum percentage error in the measurement of its density will be: (A) 4%
 - (B) 3%
 - (C) 6%
 - (D) 7%

Ans. A

Sol. $M = \pi r^2 \ell \rho$

$$\rho = \frac{\Pi}{\pi r^2 \ell}$$

$$\frac{\Delta \rho}{\rho} \times 100 = \left|\frac{\Delta m}{m}\right| \times 100 + 2\left|\frac{\Delta r}{r}\right| \times 100 + \left|\frac{\Delta \ell}{\ell}\right| \times 100$$

 $\frac{0.006}{0.6} \times 100 + 2\frac{0.005}{0.5} \times 100 + \frac{0.04}{4} \times 100$ = 1 + 2 + 1 = 4%

4. A system of two blocks of masses m = 2 kg and M = 8 kg is placed on a smooth table as shown in figure. The coefficient of static friction between two blocks is 0.5. The maximum horizontal force F that can be applied to the block of mass M so that the blocks move together will be:

F

(A) 9.8 N

- (B) 39.2 N
- (C) 49 N
- (D) 78.4 N
- Ans. (C)
- Sol. Maximum acceleration of upper block = μ g = 0.5 × 9.8 = 4.9 F_{max.} = (2 + 8) × 4.9 = 49N
- 5. Two blocks masses 10 kg and 30 kg are placed on the same straight line with coordinates (0, 0) cm and (x, 0) cm respectively. The block of 10 kg id moved on the same line through a distance of 6 cm towards the other block. The distance through which the block of 30 kg must be moved to keep the position of centre of mass of the system unchanged is:

(A) 4 cm towards the 10 kg block

- (B) 2 cm away from the 10 kg block
- (C) 2 cm towards the 10 kg block
- (D) 4 cm away from the 10 kg block

Ans. (B)

 $m_1 \Delta x + m_2 \Delta x_2 = \Delta x_{cm}$ Sol.

$$m_1 + m_2$$

 $10(6) + 30(\Delta x_2) = 0$ $\Delta x_2 + 30 (\Delta x_2) = 2 \text{ cm}$

6. A 72 Ω galvanometer is stunted by a resistance of 8 Ω . The percentage of the total current which passes through the galvanometer is:

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- (A) 0.1%
- (B) 10%
- (C) 25%
- (D) 0.25%

Ans. **(B)**

Sol.

$$i^{1_2}$$
 i_1^{0} G

$$i_1 = \left(\frac{8}{8+72}\right)i = \frac{i}{10} \rightarrow \frac{i_1}{i} \times 100 = 10\%$$

7. Given below are two statements :

Statement I: The law of gravitation holds good for any pair of bodies in the universe.

Statement II: The weight of any person becomes zero when the person is at the centre earth.

In the light of the above statements, choose the correct answer from the options given blow

- (A) Both Statement I and Statement II are true
- (B) Both Statement I and Statement II are false
- (C) Statement I is true but Statement II are false
- (D) Statement I is false but Statement II are true
- Ans. (A)
- Sol. At centre g is zero.
- 8. What percentage of kinetic energy of moving particle is transferred to a stationary parti when it strikes the stationary particle of 5 times its mass?

(Assume the collision to be head-on elastic collision)

- (A) 50.0%
- (B) 66.6%
- (C) 55.6%
- (D) 33.3%

 $v_2 = u/3$

Ans. (C)

Sol.

$$\overset{m}{\longrightarrow} \overset{u}{\longrightarrow} \overset{5m}{\longrightarrow} = \overset{m}{\longrightarrow} \overset{v_1}{\longrightarrow} \overset{m}{\longrightarrow} \overset{v_2}{\longrightarrow}$$

$$\begin{array}{c} mu + 0 = mv_1 + 5mv_2 \\ v_1 + 5v_2 = u \\ v_2 - v_1 = 1(u - 0) \\ add (1) \& (2) \\ 6v_2 = 2u \end{array}$$

$$\frac{\frac{1}{2}(5m)v_2^2}{\frac{1}{2}mv^2} \times 100 = \frac{500}{9} = 55.6$$

9. The velocity of a small ball of mass 'm' and density d1, when dropped in a container filled with a glycerine, become constant after some time. If the density of glycerine is d_2 , then the viscous force acting on the ball, will be:

(A) mg
$$\left(1-\frac{d_1}{d_2}\right)$$

(B) mg $\left(1-\frac{d_2}{d_1}\right)$
(C) mg $\left(\frac{d_1}{d_2}-1\right)$
(D) mg $\left(\frac{d_2}{d_1}-1\right)$

Sol. At the time of terminal velocity ball is in equilibrium

i.e.
$$F_g = F_b + F_v$$

 $\Rightarrow F_v = F_g - F_b$
 $\Rightarrow F_v = Mg - g$
 $\Rightarrow mg\left(\frac{d_2}{d_1} - 1\right)$

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- 10. The susceptibility of paramagnetic material is 99. The permeability of the material Wb / A-m, is: [Permeability of free space $\mu_0 = 4\pi \times 10^{-7}$ Wb / A–m]
 - (A) $4\pi \times 10^{-7}$
 - (B) $4\pi \times 10^{-4}$
 - (C) $4\pi \times 10^{-5}$
 - (D) $4\pi \times 10^{-6}$

Ans. (C)

- Sol. $\mu_r = 1 + \chi$ = 1 + 99 = 100 $\mu = \mu_r \mu_0 = 100 \times 4\pi \times 10^{-7}$ $\mu = 4\pi \times 10^{-5}$
- 11. The current flowing through an ac circuit is given by

 $I = 5 \sin(122\pi t) A$

How long will the current take to reach the peak value starting from zero? 2¢

- (A) $\frac{1}{60}$ s
- (B) 60 s
- (C) $\frac{1}{120}$ s

(D)
$$\frac{1}{240}$$
s

$$T = T = \frac{2\pi}{\omega} = \frac{2\pi}{120\pi} = \frac{1}{60}$$
$$T = \frac{1}{100}$$

 $\frac{1}{4} = \frac{1}{240}$

- Match List I with List II : 12.
 - (a) ultraviolet rays (i) study crystal structure
 - (b) Microwaves (ii) Greenhouse effect
 - (c) Infrared waves (iii) Sterilizing surgical instruments
 - (d) X-rays (iv) Radar system

Choose the correct answer from the options given below:

(A) (a)-(iii), (b)-(iv), (c)-(ii), (d)-(i)

(B) (a)-(iii), (b)-(i), (c)-(ii), (d)-(iv)

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- (C) (a)-(iv), (b)-(iii), (c)-(ii), (d)-(i)
- (D) (a)-(iii), (b)-(iv), (c)-(i), (d)-(ii)

Ans. (A)

- Sol. Theory based.
- **13.** An α particle and a carbon 12 atoms has same kinetic energy K. The ratio of their de-Broglie wavelengths (λ_{α} : λ_{C12}) is :
 - (A) 1 : √3
 - (B) √3 : 1
 - (C) 3 : 1
 - (D) 2 : √3
- Ans. (B)
- **Sol.** $\lambda = \frac{h}{\sqrt{2mK}}$
 - $\frac{\lambda_1}{\lambda_2} = \sqrt{\frac{m_2}{m_1}}$ $\frac{\lambda_1}{\lambda_2} = \sqrt{\frac{12}{4}}$
 - $\lambda_1:\lambda_2\sqrt{3}:1$
- **14.** A force of 10 N acts on a charged particle placed between two plates of a charged capacitor. If on plate of capacitor is removed, then the force acting on that particle will be.
 - (A) 5 N
 - (B) 10 N
 - (C) 20 N
 - (D) Zero

Ans. (A)

Sol. Initially force is
$$F = qE = q\left(\frac{q}{A \in_0}\right) = \frac{q^2}{A \in_0}$$

after removing one plate force is

$$F' = qE' = q\left(\frac{q}{2A \in_0}\right) = \left(\frac{q^2}{2A \in_0}\right)$$
$$\Rightarrow F' = \frac{F}{2}$$

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 $\Rightarrow F' = \frac{10}{2} = 5N$

The displacement of simple harmonic oscillator after 3 seconds starting from its mean position is equal to half of its amplitude. The time period of harmonic motion is :
 (A) 6 s

- (B) 8 s
- (C) 12 s
- (D) 36 s

Ans. (D)

Sol. $X = A \sin \omega t$

$$\frac{A}{2} = A \sin \omega t$$

$$\omega t = \frac{\pi}{6}$$

$$t = \frac{1}{12} = 3 \sec \theta$$

T = 36 sec.

16. An observe moves towards a stationary source of sound with a velocity equal to one-fifth of the velocity of sound. The percentage change in the frequency will be:

Source

- (A) 20%
- (B) 10%
- (C) 5%
- (D) 0%

Ans. (A)

Sol.

Observer C/5

$$f' = \frac{V + V_0}{V} f$$

$$f' = \frac{C + C / 5}{C} f$$

$$f' = \frac{6C}{5C} f$$

$$f' = \frac{6}{5} f$$

 $\frac{f'-f}{f} = \frac{6-5}{5}$ $\frac{\Delta f}{f} = \frac{1}{5}$ $\frac{\Delta f}{f} \times 100 = \frac{1}{5} \times 100 = 20\%$

- **17.** Consider a light ray travelling in air is incident into a medium of refractive index $\sqrt{2n}$. The incident angle is twice that a refracting angle. Then, the angle of incident will be :
 - (A) $\sin^{-1}(\sqrt{n})$ (B) $\cos^{-1}(\sqrt{\frac{n}{2}})$ (C) $\sin^{-1}(\sqrt{2n})$ (D) $2\cos^{-1}(\sqrt{\frac{n}{2}})$ (D) $1\sin 2r = \sqrt{2n} \sin r$ $2\sin r \cos r = \sqrt{2n} \sin r$ $\cos r = \sqrt{\frac{n}{2}}$ $r = \cos^{-1}\sqrt{\frac{n}{2}}$
- **18.** A hydrogen atom in its ground state absorbs 10.2 eV of energy. The angular momentum of electron of the hydrogen atom will increase by the value of:

(Given, Planck's constant = 6.6×10^{-34} Js).

(A) $2.10\times10^{-34}\,Js$

- (B) $1.05 \times 10^{-34} \text{ Js}$
- (C) 3.15×10^{-34} Js
- (D) 4.2×10^{-34} Js

Ans. (B)

Ans.

Sol.

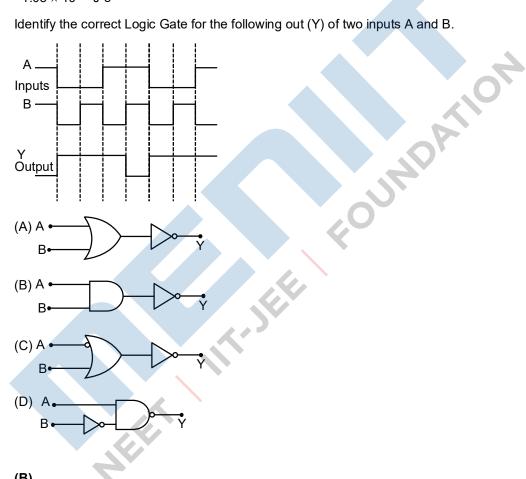
Sol.
$$\Delta E = 13.6 \times \left(\frac{1}{n_1^2} - \frac{1}{n_1^2}\right)$$

 $10.2 = 13.6 \times \left[\frac{1}{n_1^2} - \frac{1}{n_1^2}\right]$
 $n_1 = 1 \& n_2 = 2$

change in angular momentum $\Delta_L = \frac{n_2 h}{2\pi} - \frac{n_1 h}{2\pi} = \frac{h}{2\pi}$

 $=1.05 \times 10^{-34} \text{ J-s}$

19. Identify the correct Logic Gate for the following out (Y) of two inputs A and B.



Ans. (B)

Truth table Sol.

	А	В	Υ
	1	1	0
	0	0	1
	0	1	1
	1	0	1
	1	1	0
_	0	0	1

This truth table is possible in option (B).

20. A mixture of hydrogen and oxygen has volume 2000 cm³, temperature 300 K, pressure 100 kPa and mass 0.76 g. The ratio of number of moles of hydrogen to number of moles of oxygen in the mixture will be:

[Take gas constant R = $8.3 \text{ JK}^{-1}\text{mol}^{-1}$] (A) $\frac{1}{3}$ (B) $\frac{3}{1}$ (C) $\frac{1}{16}$ (D) $\frac{16}{1}$ FOUNDATIS (B) Ans. Sol. $V = 2000 \text{ cm}^3$ P = 100 kPa T = 300 K Mass of gas mixture = $0.76 \text{ g} (O_2 + H_2)$ $n = \frac{PV}{RT} = \frac{2 \times 10^{-3} \times 100 \times 10^{3} \times 3}{25 \times 300}$ JEE $n = \frac{2}{25}$ $n_2 = \frac{0.08(30 - 32 + 0.76)}{30}$ 32-0.76 0.08(32 - 0.76)n₁ $\overline{n_2}^{-}$ 0.08(30 - 32 + 0.76) (-2+0.76) $n_1(32) + n_2(2) = 0.76$ $n_2 + 16n_1 = 0.38$ $n_2 + n_1 = 0.08$ $15n_1 = 0.30$ $n_1 = \frac{0.30}{15} = 0.02$ $n_2 = 0.06$ $n_2/n_1 = 3$

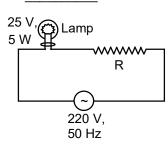
21. In a carnot engine, the temperature of reservoir is 527 °C and that of sink is 200 K. If the workdone by the engine when it transfers heat from reservoir to sink is 12000 kJ, the quantity of heat absorbed by the engine form reservoir is $\times 10^6$ J.

Sol. $\frac{W}{Q_1} = 1 = \frac{T_2}{T_1}$ $\frac{12000}{Q_1} = 1 = \frac{200}{800}$

 $Q_1 = 12000 \times 4/3 = 16000 \text{ kJ}$

22. A 220 V, 50 Hz AC source is connected to a 25 V, 5 W lamp and an additional resistance R in series (as shown in figure) to run the lamp at its peak brightness, then the value of R (in ohm) will be

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(975) Ans.

Sol.
$$i_{ms} = \frac{5}{25} = \frac{220}{R_{-} + R_{-}}$$

 $R_{B} + R = 100$

$$R_{B} = \frac{25 \times 25}{5} = 125$$

R = 1100 - 125 = 975

23. In Young's double slit experiment the two slits are 0.6 mm distance apart. Interference pattern is observed on a screen at a distance 80 cm from the slits. The first dark fringe is observed on the screen directly opposite to one of the slits. The wavelength of light will be _____ nm.

Ans. (450)

According to question Sol.

$$\frac{\beta}{2} = 0.3$$
mm $\Rightarrow \beta = 0.6$ mm

$$\Rightarrow \qquad \frac{\lambda D}{d} = 0.6mm \Rightarrow \lambda = \frac{0.6d}{D} \Rightarrow \lambda = \frac{0.6 \times 0.6}{800}$$

 $\lambda = 4.5 \times 10^{-4} \text{ mm} = 450 \text{ nm}$ \Rightarrow

24. A beam of monochromatic light is used to excite the electron in Li⁺⁺ from orbit to the third orbit. The wavelength of monochromatic light is found to be $x \times 10^{-10}$ m. The value of x is ______.

[Given hc = 1242 eV nm]

Ans. (114)

Sol.
$$\frac{hc}{\lambda} = 13.6z^2 \left[\frac{1}{n_1^2} - \frac{1}{n_2^2} \right]$$

$$\frac{1242 \times 10^{-9}}{\lambda} = 13.6 \times 9 \times \left[\frac{1}{1^2} - \frac{1}{3^2}\right]$$

 $\lambda = 114 \times 10^{-10} \text{ m}$

25. A cell, stunted by a 8 Ω resistance, is balanced across a potentiometer wire of length 3 m. The balancing length is 2 m when the cell is stunted by 4 Ω resistance. The value of interval resistance of the cell will be _____ Ω .

Sol. iR =
$$x\ell$$

$$\Rightarrow \frac{\varepsilon}{R_1 + r} R_1 = x\ell_1$$
$$\Rightarrow \frac{\varepsilon}{1 + \frac{r}{R_1}} = x\ell_1 \& \frac{\varepsilon}{1 + \frac{r}{R_2}} = x\ell_2$$

Dividing the Equation,

$$\frac{1 + \frac{r}{R_2}}{1 + \frac{r}{R_1}} = \frac{\ell_2}{\ell_1} = \frac{2}{3} \Longrightarrow 3 + \frac{3r}{8} = 2 + \frac{2r}{4} \Longrightarrow \frac{r}{2} - \frac{3r}{8} = 1$$

$$\Rightarrow$$
r = 8 Ω

26. The current density in a cylindrical wire of radius 4 mm is 4×10^6 Am⁻². The current through the outer portion of the wire between radial distances $\frac{R}{2}$ and R is ______ π A.

Ans. (48)

Sol. $i = J \times \pi \left(R^2 - \frac{R}{2} \right)$

= $4 \times 10^6 \times \pi \times 16 \times 10^{-6} \times \frac{3}{4}$

27. A capacitor of capacitance 50 pF is charged by 100 V source. It is then connected to another uncharged identical capacitor. Electrostatic energy loss in the process is ______ nJ.

Sol. Energy loss H =
$$\frac{1}{2} \frac{C_1 C_2}{C_1 + C_2} (V_1 - V_2)^2$$

$$\Rightarrow \frac{1}{2} \frac{C^2}{2C} (V - 0)^2$$
$$\Rightarrow H = \frac{1}{4} CV^2$$
$$\Rightarrow H = \frac{1}{4} \times 50 \times 10^{-12} \times 100 \times 100$$
$$\Rightarrow H = 1.25 \times 10^{-7} \text{ J} = 125 \times 10^{-9} \text{ J}$$

- **28.** The height of a transmitting antenna at the top of a tower is 25 m and that of receiving antenna is, 49 m. The maximum distance between them, for satisfactory communication in Los (Line-Of-Sight) is $K\sqrt{5} \times 10^2$ m. The value of K is _____.
- Ans. (192)
- $\textbf{Sol.} \qquad d = \sqrt{2Rh_1} + \sqrt{2Rh_2}$

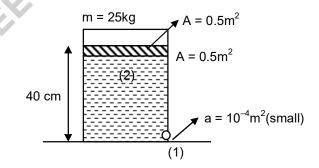
$$=\sqrt{2R}(5+7)$$

 $=\sqrt{6400 \times 2 \times 10^3}$ (12)

- $= 12\sqrt{64\times10^4\times20}$
- $= 12 \times 800 \times 2\sqrt{5} = 24 \times 800\sqrt{5} = 19200\sqrt{5}$
- $=192\sqrt{5}\times10^{2}$

K = 192

- **29.** The area of cross-section of a large tank is 0.5 m^2 . It has a narrow opening near the bottom having area of cross-section 1 cm². A load of 25 kg is applied on the water at the top in the tank. Neglecting the speed of water in the tank, the velocity of the water , coming out of the opening at the time when the height of water level in the tank is 40 cm above the bottom, will be ______ cms⁻¹. [Take g = 10 ms⁻²]
- Ans. (300)
- Sol. Apply Bernoulli equation at top and bottom



Points (1) and (2)

$$P_{0} + \frac{mg}{A} + 0 + \rho.g.h = P_{0} + \frac{1}{2}\rho u^{2} + 0$$
$$\frac{250}{5} + 10^{3} \times 10 \times 0.4 = \frac{1}{2} \times 10^{3} \times u^{2}$$

v = 3 m/s = 300 cm/s

- 30. A pendulum of length 2 m consists of a wooden bob of mass 50g. A bullet of mass 75 g is fired towards the stationary bob with a speed υ . The bullet emerges out of the bob with a speed $\frac{\upsilon}{3}$ and the bob just completes the vertical circle. The value of υ is _ ms^{-1} . (if g= 10m/s²).
- Ans. (10)
- Sol. Pendulum length 2m

$$75V = 50\sqrt{5rg} + 75(V/3)$$

$$75\left(\frac{2V}{3}\right) = 50\sqrt{5 \times 2 \times 10}$$

$$75\left(\frac{2V}{3}\right) = 50 \times 10$$

Pendulum length 2m

$$75V = 50\sqrt{5rg} + 75(V/3)$$

 $75\left(\frac{2V}{3}\right) = 50\sqrt{5 \times 2 \times 10}$
 $75\left(\frac{2V}{3}\right) = 50 \times 10$
 $V = \frac{500 \times 3}{150} = \frac{500}{50} = 10m/s$

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PART : CHEMISTRY

- 1. Given below are two statements: one is labelled as Assertion(A) and the other is labelled as Reason (R).
 - Assertion(A): At 10°C, the density of a 5 M solution of KCI [atomic masses of K & CI are 39 & 35.5 g mol⁻¹ respectively], is 'x' g ml⁻¹. The solution is cooled to –21°C. The molality of the solution will remain unchanged.
 - Reason (R): The molality of a solution does not change with temperature as mass remains unaffected with temperature.

In the light of the above statements, choose the correct answer from the options given below.

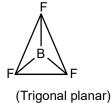
- (A) Both (A) and (R) are true and (R) is the correct explanation of (A).
- (B) Both (A) and (R) are true but (R) is not the correct explanation of (A).
- (C) (A) is true but (R) is false.
- (D) (A) is false but (R) is true.

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Ans. (A)
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- Sol. Molality & mass are temperature independent so on changing temperature molality & mass remain unchanged.
- 2. Based upon VSEPR theory, match the shape (geometry) of the molecules in list-I with the molecules in List II and select the most appropriate option.

	List – I (Shape)	(Molec			
	(A) T-shaped		(I) XeF ₄		
	(B) Trigonal planar		(II) SF4		
	(C) square planner		(III) CIF ₃		
	(D) see-saw		(IV) BF ₃		
	(A) (A) - (I), (B) - (II), (I)	C) – (III)	, (D) – (IV)		
	(B) (A) – (III), (B) – (IV)	, (C) – (I), (D) – (II)		
	(C) (A) – (III), (B) – (IV)	, (C) – (I	ll), (D) – (I)		
	(D) (A) – (IV), (B) – (III)	, (C) – (I	l), (D) – (II)		
Ans.	(B)				
Sol.	(I) BF3		(A) T-shaped		
	(II) SF4		(B) Trigonal plar	nar	
	(III) XeF ₄		(C) square plan	ner	
	(IV) CIF ₃		(D) see-saw		
	(I)	(II)		(111)	(IV)

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(See-saw)

List – II

(I) $\Delta H < 0$

(II) $\Delta G_{T,P} < 0$







(III) Isothermal and isobaric process

(T-shape)

3. Match List – I with List – II.

List – I

- (A) Spontaneous process
- (B) Process with $\Delta P = 0$, $\Delta T = 0$
- (C) $\Delta H_{reaction}$
- (D) Exothermic

(IV) [Bond energies of molecules in reactants] – [Bond energies of product molecules]

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Choose the correct answer from the options given below :

 $(A) (A) - (III), (B) - (II), (C) - (IV), (D) - (I) \\ (B) (A) - (II), (B) - (III), (C) - (IV), (D) - (I) \\ (C) (A) - (II), (B) - (III), (C) - (I), (D) - (IV) \\ (D) (A) - (II), (B) - (I), (C) - (III), (D) - (IV) \\$

Ans. (B)

- Sol. (I) For spontaneous process $\Rightarrow \Delta G < 0$
 - (II) For exothermic process $\Rightarrow \Delta H < 0$
 - (III) For isothermal process $\Rightarrow \Delta T < 0$

For isobaric process $\Rightarrow \Delta P < 0$

4. Match List – I with List – II.

List – I (A) Lyophilic colloid	List – II (I) Liquid-liquid colloid
(B) Emulsion	(II) Protective colloid
(C) Positively charged colloid	(III) FeCl₃ + NaOH
(D) Negatively charged colloid	(IV) FeCl₃ + hot water

Choose the correct answer from the options given below :

$$(A) (A) - (II), (B) - (I), (C) - (IV), (D) - (III)$$

$$(B) (A) - (III), (B) - (I), (C) - (IV), (D) - (II)$$

$$(C) (A) - (II), (B) - (I), (C) - (III), (D) - (IV)$$

(D) (A) - (III), (B) - (II), (C) - (I), (D) - (IV)

Ans. (A)

- Sol. (A) Lyophilic colloid \Rightarrow Protective colloid
 - (B) Emulsion \Rightarrow Liquid-Liquid colloid
 - (C) Positively charged colloid \Rightarrow FeCl₃ + hot water
 - (D) Negatively charged colloid \Rightarrow FeCl₃ + NaOH
- 5. Given below are two statements: one is labelled as Assertion(A) and the other is labelled as Reason (R).

Assertion(A): The ionic radii of O^{2-} and Mg^{2+} are same.

Reason (R): Both O^{2-} and Mg^{2+} are isoelectronic species.

In the light of the above statements, choose the correct answer from the options given below.

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- (A) Both (A) and (R) are true and (R) is the correct explanation of (A).
- (B) Both (A) and (R) are true but (R) is not the correct explanation of (A).
- (C) (A) is true but (R) is false.
- (D) (A) is false but (R) is true.

Ans. (D)

Sol.

lon O^{2-} Mg²⁺ No. of e⁻ 10 10

Z 8 12

Order of ionic size \Rightarrow O² > Mg²⁺

6. Match List – I with List – II.

List – I	List – II
(A) Concentration of gold ore	(I) Aniline
(B) Leaching of alumina	(II) NaOH
(C) Froth stabilizer	(III) SO ₂
(D) Blister copper	(IV) NaCN
	e

Choose the correct answer from the options given below :

(A) (A) - (IV), (B) - (III), (C) - (II), (D) - (I)

- (B) (A) (IV), (B) (II), (C) (I), (D) (III)
- (C) (A) (III), (B) (II), (C) (I), (D) (IV)

$$(D) (A) - (II), (B) - (IV), (C) - (III), (D) - (I)$$

Ans. (B)

Sol.	Theory based		
7.	Addition of H ₂ SO ₄ to BaO ₂ produces:		
	(A) BaO, SO ₂ and H ₂ O		
	(B) BaHSO₄ and O₂		
	(C) BaSO ₄ , H ₂ and O ₂		
	(D) BaSO ₄ and H ₂ O ₂		
Ans	(D)		
Sol.	$BaO_{2.}8H_{2}O(s) + H_{2}SO_{4}(aq) \longrightarrow BaSO_{4}(s) + H_{2}O_{2}(aq) + 8H_{2}O(I)$		
8.	BeCl ₂ reacts with LiAIH ₄ to give:		
	(A) Be + Li[AICl ₄] + H ₂		
	(B) Be + AlH ₃ + LiCl + HCl		
	(C) BeH ₂ + LiCl + AlCl ₃		
	(D) BeH ₂ + Li[AlCl ₄]		
Ans.	(C)		
Sol.	$2\text{BeCl}_2 + \text{LiAlC}_4 \rightarrow 2\text{BeH}_2 + \text{LiCl} + \text{AlCl}_3$		
9.	Match List – I with List – II.		
	List – I (Si Compoundo) (Si Dolumerio (Other Droducto)		
	(Si-Compounds) (Si-Polymeric/Other Products) (A) (CH ₃) ₄ Si (I) Chain Silicone		
	(B) (CH ₃)Si(OH) ₃ (II) Dimeric Silicon		
	(C) (CH ₃) ₂ Si(OH) ₂ (III) Silane		
	(D) (CH ₃) ₃ Si(OH) (IV) 2D - Silicone		
	Choose the correct answer from the options given below :		
	(A) (A) - (III), (B) - (II), (C) - (I), (D) - (IV)		
	(B) (A) – (IV), (B) – (I), (C) – (II), (D) – (III)		
	(C) (A) – (II), (B) – (I), (C) – (IV), (D) – (III)		
	(D) (A) – (III), (B) – (IV), (C) – (I), (D) – (II)		
Ans.	(D)		
Sol.	(i) Si(CH_3)_4Silane(ii) Si(CH_3)_2(OH)_2Chain silicone(iii) Si(CH_3)(OH)_32D silicone(iv) Si(CH_3)_3(OH)Dimeric silicone		
10.	Heating white phosphorus with conc. NaOH solution gives mainly:		

(A) Na₃P and H₂O

- (B) H₃PO and NaH
- (C) P(OH)3 and NaH2PO4
- (D) PH₃ and NaH₂PO₂
- Ans. (D)
- Sol. P_4 (white) + NaOH \longrightarrow PH₃ + NaH₂PO₂
- 11. Which of the following have maximum stabilization due to crystal field ?
 - (A) [Ti(H₂O)₆]³⁺
 - (B) [Co(H₂O)₆]²⁺
 - (C) [Co(CN)6]³⁻
 - (D) [Cu(NH₃)₄]²⁺
- Ans. (C)

Sol.

- ComplexHybridisation(1) $[Ti(H_2O)_6]^{3+}$ d^2sp^3

 - (2) $[Co(H_2O)_6]^{2+}$ sp³d²
 - (3) $[Co(CN)_6]^{3-}$ d²sp³
 - (4) $[Cu(NH_3)_4]^{2+}$ dsp²

 Δ_{sp} = 1.3 Δ_{0}

- 12. Given below are two statements:
 - Statement I : Classical smog occurs in cool humid climate. It is a reducing mixture of smoke, fog and sulphur dioxide.

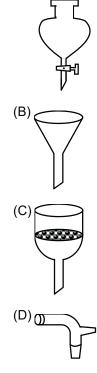
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Statement II : Photochemical smog has components, ozone, nitric oxide, acrolein, formaldehyde, PAN etc.

In the light of the above statements, choose the most appropriate answer from the options given below.

- (A) Both Statement I and Statement II are correct.
- (B) Both Statement I and Statement II are incorrect.
- (C) Statement I is correct but Statement II is incorrect.
- (D) Statement I is incorrect but Statement II is correct.
- Ans. (A)
- Sol. It is fact.
- 13. Which of the following is structure of a separating ?

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(A)

- Ans. (A)
- Sol. It is fact.
- 14. 'A' and 'B' respectively are:

 $A \xrightarrow[(2)Zn-H_2O]{(2)Zn-H_2O} E than e-1, 2-dicarbaldehy + Glyoxal/Oxoaldehyde$

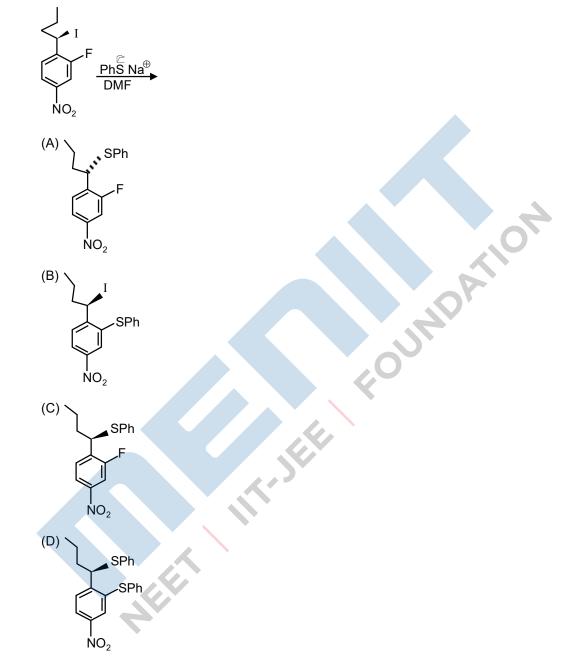
- $B \xrightarrow{(1)O_3}{(2)Zn-H_2O}$ 5-oxohexanal
- (A) 1-methylcycohex-1,3-diene & cyclopentane.
- (B) Chyclohex-1,3-diene & cyclopentane.
- (C) 1-methycycohex-1,4-diene & 1-methylcyclopent-1-ene
- (D) Cyclohex-1, 3-diene & 1-methylcyclopent-1-ene
- Ans. (D)

Sol.

 $\bigcup \frac{O^3}{Zn,H_2O}$ Ethane-1,2-Dicarbaldehyde + Glyoxal

$$rac{O^3}{Zn,H_2O}$$
 5-oxohexanal

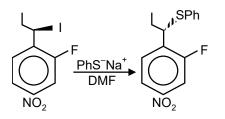
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15. The major product of the following reaction is :

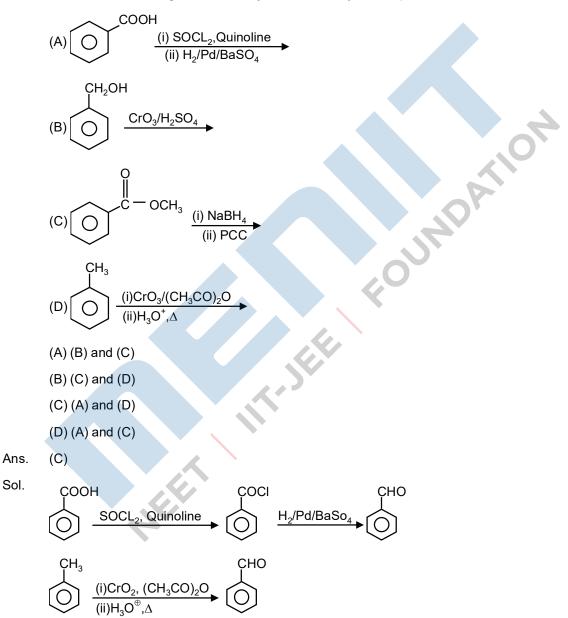
Ans. (A)

Sol. In SN² reaction on version take place





. Which of the following reactions will yield benzaldehyde as a product?



17. Given below are two statement:

Statement – I: In Hofmann degradation reaction, the migration of only an alkyl group takes place from carbonyl carbon of the amide to the nitrogen atom.

Statement – II: The group is migrate in Hofmann degradation reaction to electron deficient atom.

In the light of the above statement, choose the most appropriate answer from the options given below :

- (A) Both Statement I and Statement II are correct.
- (B) Both Statement I and Statement II are incorrect.
- (C) Statement I is correct but Statement II is incorrect.
- (D) Statement I is incorrect but Statement II is correct.
- Ans. (D) (BONUS)
- Sol. Both Statement I and statement II are incorrect statement is given in the frame of Hofmaan degradation reaction but there is no such methylmygration in Hofmaan degradation.
- 18. Match List I with List II.

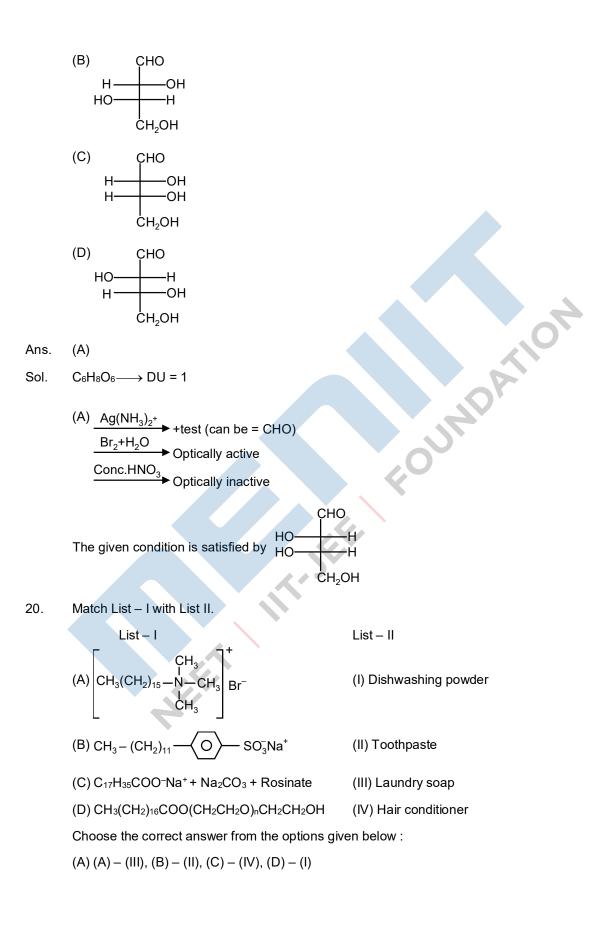
List – I (Polymer) (A) Bakelite	List – II (Used in) (I) Radio and television cabinets	
(B) Glyptal	(II) Electrical switches	
(C) PVC	(III) Paints and Lacquers	
(D) Polystyrene	(IV) Water pipes	
Choose the correct answer from the options given below :		
(A) (A) $-$ (II), (B) $-$ (III), (C) $-$ (IV), (D) $-$ (I)		
(B) (A) - (I), (B) - (II), (C) - (III), (D) - (IV)		

(C) (A) - (IV), (B) - (III), (C) - (II), (D) - (I)

$$(D) (A) - (II), (B) - (III), (C) - (I), (D) - (IV)$$

Ans. (A)

- Sol. It is fact.
- L-isomer of a compound 'A' (C₄H₈O₄) gives a positive test with [Ag(NH₃)₂]⁺. Treatment of 'A' with acetic anhydride yields triacetate derivative. Compound 'A' produces an optically active compound (B) and an optically inactive compound (C) on treatment with bromine water and HNO₃ respectively. Compound (A) is:



25

- $(B) (A) (IV), (B) (II), (C) (III), (D) (I) \\ (C) (A) (IV), (B) (III), (C) (II), (D) (I) \\ (D) (A) (III), (B) (IV), (C) (I), (D) (I)$
- Ans. (B)
- Sol. It is fact.
- 21. Metal deficiency defect is shown by Fe_{0.93}O. In the crystal, some Fe²⁺ cations are missing and loss of positive charge is compensated by the presence of Fe³⁺ ions. The percentage of Fe²⁺ ions in the Fe_{0.93}O crystal is ______. (Nearest integer)

Sol.

 $\begin{array}{c} +2 \\ x \\ x \\ (0.93 - x) \\ 2x + 3 \\ (0.93 - x) = 2 \\ 2x + 0.93 \\ x & 3 - 3x = 2 \\ 2.79 - 2 = x \\ \end{array}$ % of Fe²⁺ = $\frac{0.79}{0.93} \times 100 = 84.94\%$

Fe_{0.93}O

Ans. = 85

22. If the uncertainty in velocity and position of a minute particle in space are, 2.3×10^{-26} (ms⁻¹) and 10^{-7} (m) respectively. The mass of the particle in g is ______. (Nearest integer)

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(Given : $h = 6.626 \times 10^{-34} \text{ Js}$)

- Ans. (22)
- Sol. According to Heisenberg uncertainty Principle

$$\Delta \mathbf{x} \times \Delta \mathbf{P} \ge \frac{b}{4\pi}$$
$$\Rightarrow 10^{-7} \times \mathbf{m}.\Delta \mathbf{V} = \frac{6.62 \times 10^{-34}}{4 \times 3.14}$$

$$\Rightarrow 10^{-7} \times m \times 2.4 \times 10^{-24} = \frac{6.62 \times 10^{-34}}{4 \times 3.14}$$

 $M = 0.2196 \times 10^{-3} Kg$

=21.96 × 10⁻⁵Kg Ans. = 22

23. 2 g of none-volatile non-electrolyte solute is dissolve in 200g of two different solvents A and B whose ebullioscopic constants are in the ratio of 1 : 8. The elevation in boiling points of A and B are in the ratio $\frac{x}{y}(x : y)$. The value of y is ______. (Nearest Integer)

Ans (8)

Sol. $\Delta T_b = K_b \times m$

$$\frac{(\Delta T_{_{\rm b}})_{_{\rm I}}}{(\Delta_{_{\rm b}})_{_{\rm II}}} = \frac{(K_{_{\rm b}})_{_{\rm I}}}{(K_{_{\rm b}})_{_{\rm II}}} = \frac{1}{8} = \frac{x}{y}$$

So y = 8

24. $2NOCI(g) \rightleftharpoons 2NO(g) + CI_2(g)$

In an experiment, 2.0 moles of NOCI was placed in a one-litre flask and the concentration of NO after equilibrium established, was found to be 0.4 mol/L. The equilibrium constant at 30°C is 10^{-4} .

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Ans. (125)

 $2NOCI(g) \rightleftharpoons 2NO + CI_2$

Initially 2 At equilibrium (2

$$K_{c} = \frac{(NO)^{2}(CI_{2})}{(NOCI)^{2}} = \frac{(0.4)^{2}(0.2)}{(1.6)^{2}} = 0.0125 = 125 \times 10^{-4}$$

Ans = 125

25. The limiting molar conductivities of Nal, NaNO₃ and AgNO₃ are 12.7, 12.0 and 13.3 mS m² mol⁻¹, respectively (all at 25°C). The limiting molar conductivity of AgI at this temperature is _____mS m² mol⁻¹.

Sol. Given

 $\lambda_m^{\infty}(Nal) = 12.7Sm^2mole^{-2}$

 $\lambda_m^{\infty}(AgNO_3) = 13.3Sm^2mole^{-2}$

 $\lambda_{m}^{\infty}(NaNO_{3}) = 12Sm^{2}mole^{-2}$

$$\lambda_{m}^{\infty}(\text{Agl}) = \lambda_{m}^{\infty}(\text{AgNO}_{3}) + \lambda_{m}^{\infty}(\text{Nal}) - \lambda_{m}^{\infty}(\text{NaNO}_{3})$$

= 13.3 +12.7–12 = 14 S m²mole⁻²

26. The rate constant for a first order reaction is given by the equation :

$$lnk = 33.24 - \frac{2.0 \times 10^4 K}{T}$$

The Activation energy for the reaction is given by _____ $kJ mol^{-1}$. (In Nearest integer) (Given : R =8.3 J K⁻¹ mol⁻¹)

Sol.
$$\ell nk = 33.24 - \frac{2.0 \times 10^4}{T} K$$

$$K = Ae^{-\frac{Ea}{RT}}$$

$$nk = \ell nA - \frac{Ea}{10000 \times R} \times \frac{10^4}{T}$$

$$\operatorname{so}\left(\frac{\mathtt{L}_{a}}{10^{4} \times \mathrm{R}}\right) = 2$$

 $E_a = 2 \times 8.3 \times 10^4$

 $E_a = 166 \text{ kJ}$

- 27. The number of statement(s) correct from the following for Copper (at. No. 29) is/are
 - (A) Cu (II) complexes are always paramagnetic
 - (B) Cu (I) complexes are generally colourless
 - (C) Cu (I) is easily oxidized
 - (D) In Fehling solution, the active reagent has Cu(I)

Ans. (3)

- Sol. (A) Cu²⁺ has 3d⁹ Electronic configuration so its complexes are paramagnetic
 (B) Cu⁺¹ has 3d¹⁰ Electronic configuration so its complexes are colourless
 (C) Fehling solution contain aqueous solution of CuSO₄
 All three statement are correct.
- 28. Acidified potassium permanganate solution oxidises oxalic acid. The spin-only magnetic moment of the manganese product formed from the above reaction is ______ B.M.

(Nearest integer)

- Ans. (6)
- Sol. $2MnO_4^- + 5C_2O_4^{2-} + 16H^+ \longrightarrow 2Mn^{2+} + 10CO_2 + 8H_2O_{25}Mn^{2+} = 3d^5$ No. of unpaired electron = 5 $\mu(\text{spin only}) = \sqrt{n(n+2)}BM_{=\sqrt{5}(5+2)} = \sqrt{35}$

$$=\sqrt{3(3+2)} = \sqrt{3(3+2)} = \sqrt{3(3+2)}$$

29. Two elements A and B which from 0.15 moles of A_2B and AB_3 type compounds. If both A_2B and Ab_3 weigh equally, then the atomic weight of A is ______ times of atomic weight of B.

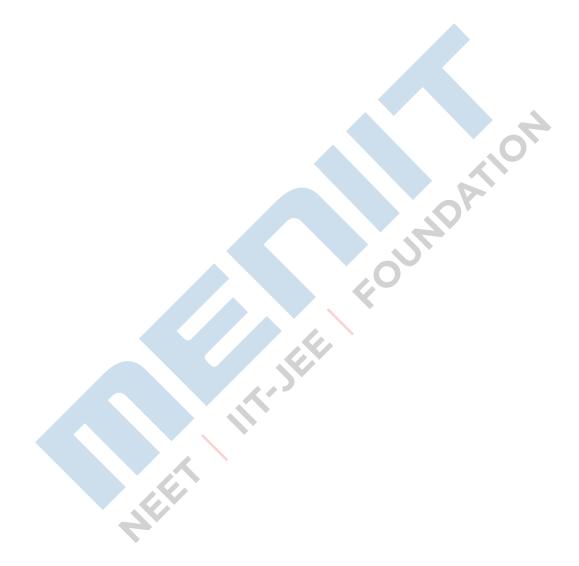
Ans. (2)

Sol. Let molar mass of A is a B is b

Again \Rightarrow 0.15 [2a + b] = 0.15 [a + 3b] a = 2b $\Rightarrow \left(\frac{a}{b}\right) = \frac{2}{1}$

30. Total number of possible stereoisomers of dimethyl cyclopentane is ______.

Ans. (6)



Part : Mathematics

SECTION-A

The area of the polygon, whose vertices are the non-real roots of the equation $\overline{z} = iz^2$ is : 1. (B) $\frac{3\sqrt{3}}{2}$ (A) $\frac{3\sqrt{3}}{4}$ (C) $\frac{3}{2}$ (D) $\frac{3}{4}$ Ans. (A) Sol. \Rightarrow Let z = x + iy, x, y \in R Now $\overline{z} = iz^2$ then $x - iy = i (x^2 - y^2 + 2xyi)$ $x - iy = i (x^2 - y^2) - 2xy$ $\Rightarrow x = -2xy \& -y = x^2 - y^2$ $\Rightarrow x(1 + 2y) = 0$ $x = 0 \text{ or } y = -\frac{1}{2}$ Put x = 0 in $- y = x^2 - y^2$ We get $y = y^2$ \Rightarrow y = 0, 1 Similarly Put $y = -\frac{1}{2}$ in $-y = x^2 - y^2$ $\Rightarrow \frac{1}{2} = x^2 - \frac{1}{4}$ $\Rightarrow x^2 = \frac{3}{4}$ $x = \pm \frac{\sqrt{3}}{2}$ $z = \left(0, i, \frac{\sqrt{3}}{2} - \frac{1}{2}i, -\frac{\sqrt{3}}{2} - \frac{1}{2}i\right)$ (0, 1)

 $\left(\frac{\sqrt{3}}{2},-\frac{1}{2}\right)$

 $\left(-\frac{\sqrt{3}}{2},-\frac{1}{2}\right)$

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Area
$$= \frac{1}{2} \cdot \left(\sqrt{3}\right) \left(\frac{3}{2}\right)$$
$$= \frac{3\sqrt{3}}{4}$$

2. Let the system of linear equation x + 2y + z = 2, $\alpha x = 3y - z = \alpha$, $-\alpha x + y + 2z = -\alpha$ be inconsistent Then α is equal to :

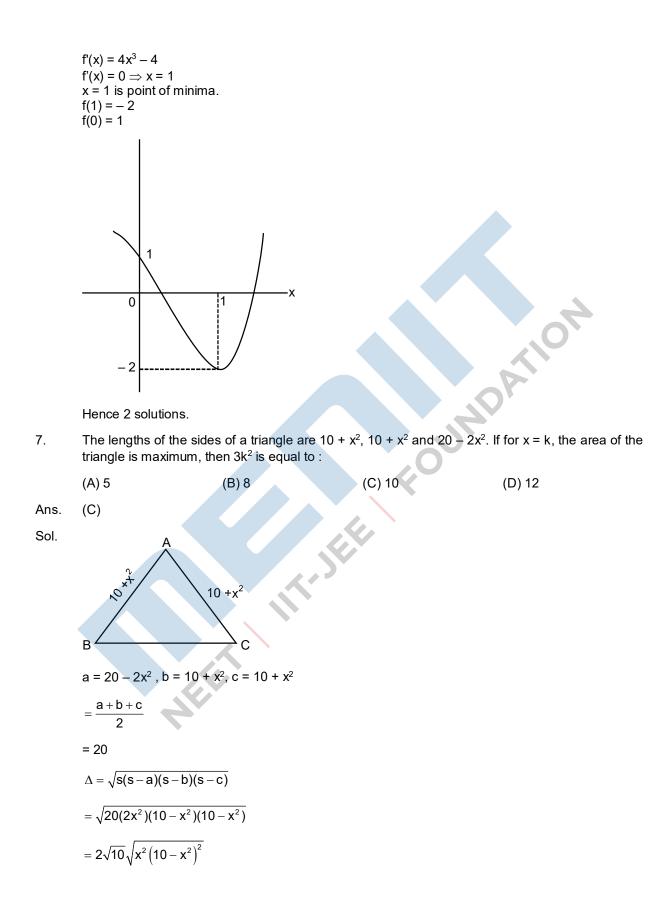
(A)
$$\frac{5}{2}$$
 (B) $-\frac{5}{2}$ (C) $\frac{7}{2}$ (D) $-\frac{7}{2}$
Ans. (D)
Sol. $\Delta = \begin{vmatrix} 1 & 2 & 1 \\ 2 & 3 & -1 \\ -2 & 1 & 2 \end{vmatrix}$
 $= (6 + y) - 2(2\alpha - \alpha) + 1(\alpha + 3\alpha)$
 $= 7 - 2\alpha + 4\alpha$
 $= 7 + 2\alpha$
 $\Delta = 0 \Rightarrow \alpha = -\frac{7}{2}$
 $\Delta_1 = \begin{vmatrix} 2 & 2 & 1 \\ -\alpha & 1 & 2 \end{vmatrix}$
 $= 14 + 2\alpha$
 $\alpha = -x_2 = 7$
 $\Delta_1 \neq 0$
3. If $x = \sum_{n=0}^{\infty} a^n, y = \sum_{m=0}^{\infty} b^n, z = \sum_{n=0}^{\infty} c^n$, where a, b, c are ibn A.P. and $|a| < 1, |b| < 1, |c| < 1$, $abc \neq 0$, then
(A) x, y, z are in A.P.
(B) x, y, z are in A.P.
(D) $\frac{1}{x} + \frac{1}{y} + \frac{1}{z} = 1 - (a + b + c)$
Ans. (C)
Sol. $x = 1 + a + a^2 =$

4.

(D) 5

 $x = \frac{1}{1-a} \Rightarrow a = 1-\frac{1}{x}$ $y = \frac{1}{1-b} \Rightarrow b = 1-\frac{1}{v}$ $z = \frac{1}{1-c} \Rightarrow c = 1-\frac{1}{z}$ a, b, c are in A.P. $\Rightarrow 1 - \frac{1}{x}, 1 - \frac{1}{y}, 1 - \frac{1}{z}$ are in A.P. $\Rightarrow -\frac{1}{x}, -\frac{1}{y}, -\frac{1}{z}$ are in A.P. $\Rightarrow \frac{1}{x}, \frac{1}{v}, \frac{1}{z}$ are in A.P. Let $\frac{dy}{dx} = \frac{ax - by + a}{bx + cy + a}$, where a, b, c are constants, represent a circle passing through the point (2, 5). Then the shortest distance of the point (11, 6) from this circle is : (A) 10 (B) 8 (C) 7 (B) Ans. Sol. Let equation of circle is THE $x^{2} + y^{2} + 2gx + 2fy + c = 0$ $\Rightarrow \frac{dy}{dx} = \frac{-(2x+2g)}{(2y+2f)}$ Comparing with $\frac{dy}{dx} = \frac{ax - by + a}{bx + cy + a}$ ⇒ b = 0, a = –2, c = 2 $\Rightarrow -2g = -2 \Rightarrow g = 1 \qquad 2f = -2$ F = - 1 Now circle will be $x^{2} + y^{2} + 2x + 2y + c = 0$ its passes through (2, 5) which will give c = -23so circle will be $x^2 + y^2 + 2x - 2y - 23 = 0$ centre C = (-1, 1)and radius 5

Now P is (11, 6) So minimum distance of P from circle will be $=\sqrt{(11+1)^2+(6-1)^2}-5$ = 13 – 5 = 8 Let a be an integer such that $\lim_{x\to 7} \frac{18 - [1 - x]}{[x - 3a]}$ exists, where [t] is greatest integer \leq t. Then a is equal 5. to : (A) - 6(B) – 2 (C) 2 (D) 6 (A) Ans. $\lim_{x \to 7} \frac{18 - [1 - x]}{[x] - 3a}$ Sol. L.H.L. $\lim_{x \to 7} \frac{18 - [1 - x]}{[x] - 3a}$ $=rac{18-(-6)}{6-3a}$ $=\frac{24}{6-3a}$ R.H.L. $\lim_{x \to 7^+} \frac{18 - [1 - x]}{[x] - 3a}$ $=\frac{18-(-7)}{7-3a}$ $=\frac{25}{7-3a}$ Now L.H.L. = R.H.L. $=\frac{24}{6-3a}=\frac{25}{7-3a}$ ⇒ 168 – 72a = 150 – 75a ⇒ 18 = – 3a \Rightarrow a = -66. The number of distinct real roots of $x^4 - 4x + 1 = 0$ is : (A) 4 (B) 2 (D) 0 (C) 1 Ans. (B) Sol. Let $f(x) = x^4 - 4x + 1$



 $=2\sqrt{10}\left|x\left(10-x^{2}\right)\right|$ $= 2\sqrt{10} \left| (10x - x^3) \right|$ $S = 10x - x^3$ $\frac{ds}{dx} = 10 - 3x^2$ $\frac{ds}{dx} = 0 \Longrightarrow x^2 \frac{10}{3}$ $3x^2 = 10$ FOUNDATION If $\cos^{-1}\left(\frac{y}{2}\right) = \log_{e}\left(\frac{x}{5}\right)^{5}$, |y| < 2, then : 8. (A) $x^2 y'' + xy' - 25y = 0$ (B) $x^2 y'' - xy' - 25y = 0$ (C) $x^2 y'' - xy' + 25y = 0$ (D) $x^2 y'' + xy' + 25y = 0$ Ans. (D) $\cos^{-1}\left(\frac{y}{2}\right) = \log_{e}\left(\frac{x}{5}\right)^{5}$ Sol. $\cos^{-1}\left(\frac{y}{2}\right) = 5\log_{e}\left(\frac{x}{5}\right)$ $\frac{-1}{\sqrt{1-\frac{y^2}{4}}} \cdot \frac{y'}{2} = 5 \cdot \frac{1}{\frac{x}{5}} \times \frac{1}{5}$ $\Rightarrow \frac{-y'}{\sqrt{4-y^2}} = \frac{5}{x}$ $-xy' = 5\sqrt{4-y^2}$ $-xy''-y'=5.\frac{1}{2\sqrt{4-y^2}}(-2yy')$ \Rightarrow xy"+ y' = $\frac{5y'.y}{\sqrt{4-y^2}}$ $xy''+y'=5.\left(\frac{-5}{x}\right)y$ $X^2y'' + xy' = -25y$

9.
$$\int \frac{(x^2 + 1)e^x}{(x + 1)^2} dx = f(x)e^x + C, \text{ Where C is a constant, then } \frac{d^2f}{dx^3} \text{ at } x = 1 \text{ is equal to :} \\ (A) - \frac{3}{4} (B) \frac{3}{4} (C) - \frac{3}{2} (D) \frac{3}{2} \\ \text{Ans. (B)} \\ \text{Sol. } \int \left(\frac{x^2 + 1}{(x + 1)^2}\right)e^x dx \\ = \int \left(\frac{x^2 - 1 + 2}{(x + 1)^2}\right)e^x dx \\ = \int \left(\frac{x - 1}{(x + 1)^2}\right)e^x dx \\ = \int \left(\frac{x - 1}{x + 1} + \frac{2}{(x + 1)}\right)e^x dx \\ = f(x)e^x + c \\ \text{Where f } (x) = \frac{x - 1}{x + 1} \\ f'(x) \frac{2}{(x + 1)^2} \\ = \frac{12}{(x + 1)^4} \\ f''(1) = \frac{12}{16} \\ = \frac{3}{4} \\ 10. \quad \text{The value of the integral } \int_{-2}^{2} \frac{|x^3 + x|}{(e^{|x|} + 1)} dx \text{ is equal to :} \\ (A) 5e^2 (B) 3e^{-2} (C) 4 (D) 6 \\ \text{Ans. (D)} \\ \text{Sol. } f(x) = \frac{|x^3 + x|}{(e^{|x|} + 1)} dx \\ \end{cases}$$

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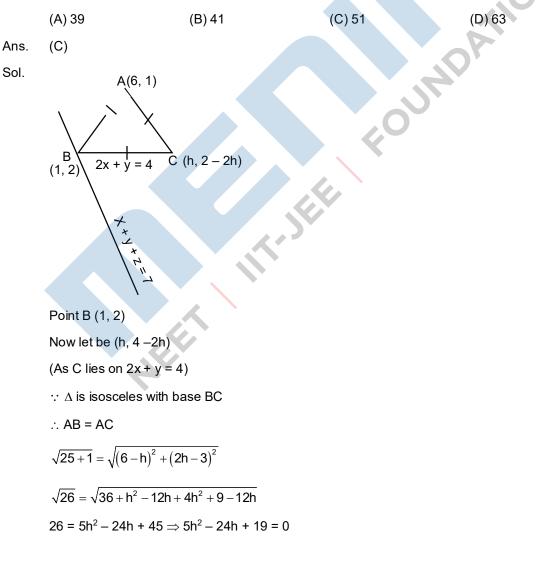
$$\begin{split} & \int_{-2}^{2} f(x) dx = \int_{0}^{2} (f(x) + f(-x)) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{iy|} + 1)} + \frac{|-x^{3} - x|}{(e^{-iy|} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{iy|} + 1)} + \frac{|x^{3} + x|}{(e^{-iy|} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{ix} + 1)} + \frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} + \frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{-x} + 1)} + \frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{-x} + 1)} + \frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(e^{-x} + 1)} + \frac{|x^{3} + x|}{(e^{-x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + e^{-x^{2}})} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(1 + e^{x^{2}} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + 1)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{2}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{3}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{3}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{3}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{3}}(x^{3} + x)}{(x^{3} + x)} \right) dx \\ & = \int_{0}^{2} \left(\frac{|x^{3} + x|}{(x^{3} + x)} + \frac{e^{x^{3}}$$

Putting this values in above relation we get C = 0

$$ln|2^{y} - 1| + ln|2^{x} - 1| = 0$$

(2^x - 1)(2^y - 1) = 1
2^y - 1 = $\frac{1}{2^{x} + 1}$
At x = 2
2^y $\frac{1}{3} + 1 = \frac{4}{3}$
y = $log_{2} \frac{4}{3} = log_{2} 4 log_{2} 3 = 2 - log_{2} 3$

12. In a isosceles triangle ABC, the vertex A is (6, 1) and the equation of the base BC is 2x + y = 4. Let the point B lie on the line x + 3y = 7. If (α, β) is the centroid $\triangle ABC$, then 15 $(\alpha + \beta)$ is equal to :



$$\Rightarrow 5h^{2} - 5h - 19h + 19 = 0$$

$$h = \frac{19}{5} \text{ or } h = 1$$
Thus $C\left(\frac{19}{5}, -\frac{18}{5}\right)$
Centroid $\left(\frac{6+1+\frac{19}{3}}{3}, \frac{1+2-\frac{18}{5}}{3}\right)$
 $\left(\frac{35+19}{15}, \frac{15-18}{15}\right)$
 $\left(\frac{54-3}{15}, \frac{15}{15}\right)$
 $a = \frac{54}{15}, \beta = \frac{-3}{15}$
15 $(\alpha + \beta) = 51$
13. Let the eccentricity of an ellipse $\frac{x^{2}}{a^{2}} + \frac{y^{2}}{b^{2}} = 1 a > b$, be $\frac{1}{4}$. If this ellipse passes through the point $\left(-4\sqrt{\frac{2}{5}}, 3\right)$, then $a^{2} + b^{2}$ is equal to:
(A) 29 (B) 31 (C) 32 (D) 34
Ans. (B)
Sol. $\frac{x^{2}}{a^{2}} + \frac{y^{2}}{b^{2}} = 1a > b$
 $e^{2} = 1 - \frac{b^{2}}{a^{2}}$
 $\frac{1}{16} = 1 - \frac{b^{2}}{a^{2}}$
 $\frac{1}{a^{2}} = 1 - \frac{16}{16} = \frac{15}{16} \Rightarrow b^{2} = \frac{15}{16} a^{2}$
 $\frac{x^{2}}{a^{2}} + \frac{y^{2}}{b^{2}} = 1$

$$\begin{aligned} \frac{32}{5a^2} + \frac{9}{15a^3} = 1 \\ \frac{80}{5a^2} = 1 \\ 16 = a^2 \\ b^2 = 15 \end{aligned}$$
14. If two straight lines whose direction cosines are given by the relations $I + m - n = 0$, $3I^2 + m^2 + cnI = 0$ are parallel, then the positive value of c is :
(A) 6 (B) 4 (C) 3 (D) 2 \\
Ans. (A) \\
Sol. I + m - n = 0 \\ 3I^2 + m^2 + cI + (I + m) = 0 \\ n = I + m \\ 3I^2 + m^2 + cI + (I + m) = 0 \\ (3 + c)^2 + clm + m^2 = 0 \\ (3 + c)^2 + clm + m^2 = 0 \\ (3 + c)^2 + clm + m^2 = 0 \\ (2 + c) (\frac{1}{m})^2 + c(\frac{1}{m}) + 1 = 0 \\ \cdots (1) \\
\because \text{ lies are parallel.} \\
\text{Roots of (1) must be equal } \\
\Rightarrow D = 0 \\ c^2 - 4(3 + c) = 0 \\ c^2 - 6(3 + c)

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 \Rightarrow \vec{a} is perpendicular to \vec{b} as well as \vec{a} is perpendicular to \vec{c}

Now $\vec{a}.\vec{c} = 2 - 3 - 2 = -3 \neq 0$

This $\vec{b} \times \vec{c} = \vec{a}$ is not possible.

No. of vectors $\vec{b} = 0$

16. Five numbers x₁, x₂, x₃, x₄, x₅ are randomly selected from the numbers 1, 2, 3,, 18 and are arranged in the increasing order ($x_1 < x_2 < x_3 < x_4 < x_5$). The probability that $x_2 = 7$ and $x_4 = 11$ is : (D) $\frac{1}{34}$

(A)
$$\frac{1}{136}$$
 (B) $\frac{1}{72}$ (C) $\frac{1}{68}$

7 16

(B)

Ans. (C)

- Sol. No. of ways to select and arrange x₁, x₂, x₃, x₄, x₅ from 1, 2, 3......18
 - $n(s) = {}^{18}C_5$

$$n(E) = {}^{6}C_{1} \times {}^{3}C_{1} \times {}^{7}C_{1}$$

$$P(E) = \frac{6 \times 3 \times 7}{{}^{18}C_5}$$
$$\frac{1}{10} = \frac{1}{10}$$

$$\overline{17 \times 4} = \overline{68}$$

- Let X be a random variable having binomial distribution B(7, p). If P(X = 3) = 5P(X = 4), then the 17. sum of the mean and the variance of X is : (C) $\frac{77}{36}$ (D) $\frac{49}{16}$
 - 105 (A) 16

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Ans. (C)

Sol. B(7, p)

N = 7 p = p

given

$$P(x = 3) = 5P(x = 4)$$

$$^{7}C_{3} \times p^{3} (1 - p)^{4} = 5^{7} \cdot C^{4} (1 - p)^{3}$$

$$\frac{^{7}C_{3}}{5 \times ^{7}C_{4}} = \frac{p}{1 - p}$$

$$1 - p = 5p$$

6p = 1

$$\frac{p}{1-p}p = \frac{1}{6} \Longrightarrow q = \frac{5}{6}$$

n = 7 Mean = np = $7 \times \frac{1}{6} = \frac{7}{6}$ Var = npq = $7 \times \frac{1}{6} = \frac{5}{6} = \frac{35}{36}$ Sum $=\frac{7}{6}+\frac{35}{36}$ $=\frac{42+35}{36}$ $=\frac{77}{36}$ The value of $\cos\left(\frac{2\pi}{7}\right) + \cos\left(\frac{4\pi}{7}\right) + \cos\left(\frac{6\pi}{7}\right)$ is equal to : 18. **OUN** (B) $-\frac{1}{2}$ $(C) - \frac{1}{3}$ 1 4 (D) (A) – 1 Ans. (B) $\cos\frac{2\pi}{7} + \cos\frac{4\pi}{7} + \cos\frac{6\pi}{7}$ Sol. $=\frac{\sin\left(3\times\frac{\pi}{7}\right)}{\sin\frac{\pi}{7}}\times\cos\left(\frac{\frac{2\pi}{7}+\frac{6\pi}{7}}{2}\right)$ $=\frac{2\sin\left(\frac{3\pi}{7}\right)}{2\sin\frac{\pi}{7}}\times\cos\left(\frac{4\pi}{7}\right)$ $=\frac{2\sin\left(\frac{7\pi}{7}\right)+\sin\left(\frac{7\pi}{7}\right)}{1+\sin\left(\frac{7\pi}{7}\right)}$ $2\sin\frac{\pi}{\pi}$ $=\frac{-\sin\frac{\pi}{7}}{2\sin\frac{\pi}{7}}$ $=-\frac{1}{2}$

Ans. (99)

Sol.
$$f(x) + f(1-x) = \frac{2e^{2x}}{e^{2x} + e} + \frac{2e^{2-2x}}{e^{2-ex} + e} \left[\frac{e^{2x}}{e^{2x} + e} + \frac{e^{2}}{e^{2} + e^{2x+1}} \right]$$
$$= 2 \left[\frac{e^{2x-1}}{e^{2x-1} + 1} + \frac{1}{1 + e^{2x-1}} \right] = 2$$
$$f\left(\frac{1}{100}\right) + f\left(\frac{2}{100}\right) + f\left(\frac{3}{100}\right) + \dots + f\left(\frac{99}{100}\right)$$
$$= \left\{ f\left(\frac{1}{100}\right) + f\left(\frac{99}{100}\right) \right\} + \left\{ f\left(\frac{2}{100}\right) + f\left(\frac{98}{100}\right) \right\} + \dots + f\left\{ \left(\frac{49}{100}\right) + f\left(\frac{51}{100}\right) \right\} + f\left(\frac{1}{2}\right)$$
$$= (2 + 2 + 2 - \dots - 49 \text{ times}) + \frac{2e}{e+e}$$
$$= 98 + 1 = 99$$
2. If the sum of all the roots of the equation $e^{2x} - 11e^{x} - 45e^{-x} + \frac{81}{2} = 0$ is loge P, is equal Ans. (45)
Sol. $e^{2x} - 11e^{x} - 45e^{-x} + \frac{81}{2} = 0$]
$$e^{x} = t$$
$$2t^{3} - 22t^{2} + 81t - 90 = 0$$
$$t_{1}t_{2}t_{3} = 45$$
$$e^{x_{1}} \cdot e^{x_{2}} \cdot e^{x_{3}} = 45$$
$$\log e^{x_{1}} \cdot x_{2} + x_{3} = \log 45$$

$$(e^{x})^{3} - 11(e^{x})^{2} - 45 + \frac{6}{2}$$

$$e^{x} = t$$

$$2t^{3} - 22t^{2} + 81t - 90 = 0$$

$$t_{1} t_{2} t_{3} = 45$$

$$e^{x}_{1} \cdot e^{x}_{2} \cdot e^{x}_{3} = 45$$

$$e^{x}_{1} + x_{2} + x_{3} = 45$$

$$\log_{e} e^{x}_{1} + x_{2} + x_{3} = \log_{e} 45$$

$$\log_{e} P = \log_{e} 45$$

$$\log_{e} P = \log_{e} 45$$

14 4

The positive value of the determinant of the matrix A, whose $Adj(Adj(A)) = \begin{pmatrix} 14 & 28 & -14 \\ -14 & 14 & 28 \\ 28 & -14 & 14 \end{pmatrix}$ is 3.

Ans. (14)
Sol.
$$\operatorname{Adj}(\operatorname{Adj}(A)) = \begin{bmatrix} 14 & 18 & -14 \\ -14 & 14 & 28 \\ 28 & -14 & 14 \end{bmatrix}$$

 $|\operatorname{Adj}(\operatorname{Adj}A)| = \begin{bmatrix} 14 & 18 & -14 \\ -14 & 14 & 28 \\ 28 & -14 & 14 \end{bmatrix} = 14 \times 14 \times 14 \begin{vmatrix} 1 & 2 & -1 \\ -1 & 1 & 2 \\ 2 & -1 & 1 \end{vmatrix}$
 $= (14)^3 [3 - 2(-5) - 1(-1)] = (14)^3 [14] = (14)^4$

 $|\mathsf{A}|^4 = (14)^4 \Longrightarrow |\mathsf{A}| = 14$ 4. The number of ways, 16 identical cubes, of which 11 are blue and rest are red, can be placed in a row so that between any two red cubes there should be at least 2 blue cubes, is Ans. (56)Sol. 11 Blue 16 cubes < 5 Red $x_1 + x_2 + x_3 + x_4 + x_5 + x_6 = 11$ $x_1, x_6 \ge 0,$ $x_2, x_3, x_4, x_5 \ge 2$ $x_2 = t_1 + 2$ $x_3 = t_3 + 2$ $x_4 = t_4 + 2$ $x_5 = t_5 + 2$ x₁, t₂, t₃, t₄, t₅, x₆ \ge 0 No. of solutions =⁶⁺³⁻¹ C₃ =⁸ C₃ = 56 \sqrt{x} $\sqrt{5}$ 5. If the coefficient of x^{10} in the binomial expansion pf is 5^{k} I, where I, $k \in N$ and I is co-OUNDAT prime to 5, then k is equal to Ans. (5) $\left(\frac{\sqrt{x}}{\frac{1}{54}} + \frac{\sqrt{5}}{\frac{1}{54}}\right)$ Sol. $T_{r+1} = {}^{60} C_r \left(\frac{x^{\frac{1}{2}}}{z^{\frac{1}{4}}} \right)^{60-r} \left(\frac{5^{\frac{1}{2}}}{z^{\frac{1}{4}}} \right) r$ JEE $=^{60} C_r 5 \frac{3r - 60}{4} . x \frac{180 - 5r}{6}$ $\frac{180-5r}{6}=10 \Rightarrow r=24$ Coeff. of $x^{10} = {}^{60} C_{24} \cdot 5^3 = \frac{|\underline{60}|}{|\underline{24}|36} 5^3$ Powers of 5 in $={}^{60}$ C₂₄.5³ $=\frac{5^{14}}{5^4 \times 5^8} \times 5^3 = 5^5$ Let $A_1 = \{(x, y) : |x| \le y^2, |x| + 2y \le 8\}$ and $A_2 = \{(x, y) : |x| + |y| \le k\}$. If 27 (Area A_1) = 5 (Area A_2), 6. then k is equal to : Ans. (6)Sol. $A_1 = \{(x, y) : |x| \le y^2, |x| + 2y \le 8\}$ and $A_2 = \{(x, y) : |x| + |y| \le k\}$. $v^2 = x$ x + 2y = 8

$$area(A_{1}) = 2\left[\frac{2}{0}y^{2}dy + \frac{4}{2}(8-2y)dy\right]$$

$$= 2\left[\left(\frac{y^{3}}{3}\right)_{0}^{2} + (8y-y^{2})_{2}^{4}\right]$$

$$area(A_{1}) = 2 \times \frac{20}{3} = \frac{40}{3}$$

$$area(A_{1}) = 2 \times \frac{20}{3} = \frac{40}{3}$$

$$Area(A_{2}) = 4 \times \frac{1}{2}k^{2}$$

$$Area(A_{2}) = 4 \times \frac{1}{2}k^{2}$$

$$Area(A_{2}) = 2k^{2}$$

$$Now$$

$$27 (Area A_{1}) = 5 (Area A_{2})$$

$$9 \times 4 = k^{2}$$

$$K = 6$$
7. If the sum of the first ten terms of the series $\frac{1}{5} + \frac{2}{65} + \frac{3}{325} + \frac{4}{1025} + \frac{5}{2501} + \dots$ is $\frac{m}{n}$, where m and n are co-prime numbers, then m + n is equal to
$$Ars. (276)$$
Sol. $\frac{1}{5} + \frac{2}{65} + \frac{3}{325} + \frac{4}{1025} + \frac{5}{2501} + \dots$

$$T_{n} = \frac{n}{4n^{n} + 1} = \frac{n}{(2n^{2} + 1)^{2} - (2n)^{2}} = \frac{n}{(2n^{2} + 2n + 1)(2n^{2} - 2n + 1)}$$

$$= \frac{1}{4} \left[\frac{1}{2n^{2} - 2n + 1} - \frac{1}{2n^{2} + 2n + 1} \right]$$

$$S_{10} = \sum_{n=1}^{m} T_{n} = \frac{1}{4} \left[\frac{1}{4} - \frac{1}{5} + \frac{1}{5} - \frac{1}{13} + \dots + \frac{1}{200 + 20 + 1} \right]$$

$$=\frac{1}{4}\left[1-\frac{1}{221}\right]=\frac{1}{4}\times\frac{220}{221}-\frac{55}{221}=\frac{m}{n}$$

m + n = 55 + 221 = 276 A rectangle R with end points of the one of its dies as (1, 2) and (3, 6) is inscribed in a circle. If the 8. equation of a diameter of the circle is 2x - y + 4 = 0, then the area of R is _____. (16)

Ans.

$$\left(\frac{1}{4}, -\frac{\sqrt{63}}{4}\right)$$
 can't touch given parabola.
Equation of circle is

$$\left(x - \frac{1}{4}\right)^{2} + \left(k - \frac{\sqrt{63}}{4}\right)^{2} = 4$$
From figure
 $\alpha = 2 + \frac{\sqrt{63}}{4} = \frac{8 + \sqrt{63}}{4}$
 $4\alpha - 8 = \sqrt{63}$
 $4\alpha - 8 = \sqrt{63}$
 $4\alpha - 8 = \sqrt{63}$
($4\alpha - 8\right)^{2} = 63$
10 Let the mirror image of the point (a, b, c) with respect to the plane $3x - 4y + 12z + 19 = 0$ be (a - 6, β, γ). If $a + b + c = 5$, then $7\beta - 9\gamma$ is equal to ______.
Ans. (137)
Sol.

$$P(a, b, c)$$

$$P(a, b, c) \rightarrow D.R 6, b - \beta, c - \gamma$$

$$M = \left(a - 3, \frac{\beta + b}{2}, \frac{\gamma + c}{2}\right)$$
Since M lies on $3x + 4y + 12z + 19 = 0$
 $\Rightarrow 6a - 4b + 12c - 4\beta + 12\gamma + 20 = 0$
 $\Rightarrow 6a - 4b + 12c - 4\beta + 12\gamma + 20 = 0$
 $\Rightarrow 6a - 4b + 12c - 4\beta + 12\gamma + 24 = 5$
 $\Rightarrow a = -\beta - \gamma - 11$
Now putting these values in (1) we get
 $6(-\beta - \gamma - 11) - 4(\beta - 8) + 12(\gamma + 24) - 4\beta + 12\gamma + 20 = 0$
 $\Rightarrow 7\beta - 9\gamma = 170 - 33 = 137$

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